



A STUDY ON ELECTROMAGNETIC PROPERTIES OF NANOMATERIALS

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Abstract - We have arranged nanocomposites comprising of barely measured metal-containing nanoparticles implanted in a polyethylene network and have laid out conditions for the manufacture of thick movies and mass materials from the blended polymer powders. Dielectric permittivity and resistivity estimations show that the electrical properties of the nanocomposites rely essentially upon the nanoparticle size and content. The microwave retention and permittivity of the materials are displayed to differ minimal in a wide recurrence range. The charge (counting the remanent one) of the cobalt-containing nanomaterials is higher than that of the iron-containing tests.

Key Words: Electromagnetic, nanomaterials, polymer, polyethylene.

1 INTRODUCTION

The investigation of metal-containing nanoparticles, incorporating those balanced out in polymer networks, is animated by the steadily expanding revenue in nanotechnology in numerous areas of science, physical science, and materials research^{1,2}. The chance of making materials that join properties of polymers and metals, and ways to deal with fitting their properties through compositional control have for quite some time been talked about in the writing⁴. Around here of examination, a few phases can be recognized. In the first place, the most consideration was paid to the improvement of advantageous and reproducible cycles for the blend of metal-containing nanoparticles, their streamlining, and amiability to commercialscale creation.^{5,6} Later endeavors were centered around compelling systems for settling metal nanoparticles³.

It is notable that most materials in view of metal containing nanoparticles are thermodynamically shaky. They can be settled utilizing various polymers, e.g., polyethylene, polypropylene, poly(tetrafluoroethylene), and others^{7,8}. These polymers offer somewhat high warm dependability, exceptional rheological properties, and high dielectric strength. Moreover, they are synthetically dormant and have good handling properties, appropriate for the creation of complex-molded and huge parts. Additionally significant is that there are experienced techniques for the combination of these polymers⁹. Generally speaking, polymeric materials are great dielectrics with stable physical and substance properties.

Attributable to their compound strength, polymers can be utilized under extreme circumstances however are utilized principally as protectors. Note that the mechanical properties of polymers can be tuned utilizing different inorganic fillers, which were accounted for to impact not just the mechanical and thermomechanical properties of polymers yet additionally their electrical qualities. Composites delivered by changing polymer

dielectrics with carbon nanotubes or metal-containing fillers have higher electrical conductivity in examination with the parent polymer framework¹⁰. As displayed in the previously mentioned examinations, the electrical properties of composites rely upon the arrangement, shape, size, and convergence of filler particles. It is additionally trusted that changing the convergence of nanoparticles in polymers, one have some control over the electrical properties of the resultant nanocomposites¹¹.

Note that the above properties rely straightforwardly upon the fixation, shape, and surface movement of filler particles. Nanometer-sized fillers, especially nanoparticles, are the most alluring for the manufacture of metal-containing polymer-framework nanocomposites on the grounds that they have various novel qualities missing in mass materials¹².

The investigation of polymer-lattice nanocomposites is vital in light of the fact that such nanosystems offer an extraordinary mix of substance, physical, mechanical, and handling properties: expanded warm and electrical conductivity, high attractive helplessness, and the capacity to retain ionizing radiation. Metal-containing polymeric materials have been the subject of serious investigations since they have impressive potential for the majority innovative applications. Metal-polymer composites can be ready by different techniques: openness of polymer movies to metal fume, substance responses of metal salts in polymer arrangements followed by the recuperation of the polymer, polymerization of different metal-containing monomeric frameworks, and others^{13,14}.

Stacking polymers with inorganic fillers offers the chance of making materials that consolidate the properties of the filler and framework. The polymer the most ideal to the application region being referred to is high-pressure polyethylene since it is similarly economical, and there is developed innovation for its creation. One more significant benefit of polyethylene is that it can undoubtedly be blended in with both natural and inorganic fillers¹⁵. Besides, polyethylene is a thermoplastic polymer, which empowers the creation of complex-molded and huge parts under gentle circumstances. Attributable to these worthwhile properties, polyethylene is broadly utilized in electrical designing¹⁶.

The utilization of such composites as circulated nonlinear parts in microwave designing (to fill waveguide lines and cavities, to deliver slender coatings and electromagnetic safeguards, and so on) is supposed to finish in a variety of novel gadgets for electromagnetic sign change and covertness applications. Polyethylene, specifically, high-sub-atomic polyethylene, is being utilized progressively as a critical part of prosthetic materials^{17,18}. Also, the high electric strength and resistivity of polyethylene make it an appealing essential part of assorted protecting materials. The above features that the improvement of powerful methodologies for the manufacture of polyethylenematrix composites is a basic issue in composite materials research¹⁹. In this paper, we report the combination and portrayal of nanocomposites in which high-pressure polyethylene is utilized as a nanoparticle-settling lattice, and the filler comprises of iron-or cobalt-containing nanoparticles of controlled size and piece.

2 EXPERIMENTAL

Tests comprising of iron-and cobalt-containing nanoparticles settled in a polyethylene network were ready through warm disintegration of metal-containing forerunners in a high-temperature polyethylene-oil arrangement at 270-300^{17,18}. The metal-containing forerunners utilized in this study were iron(III) formate, $\text{Fe}(\zeta\text{COO})_3 \cdot 2\text{H}_2\text{O}$; iron pentacarbonyl, $\text{Fe}(\text{CO})_5$; iron(II) oxalate, $\text{FeC}_2\text{O}_4 \cdot 2\text{H}_2\text{O}$; and cobalt acetic acid derivation, $(\text{CH}_3\text{COO})_2 \cdot 4\text{H}_2\text{O}$. A suitable measure of a metal-containing forerunner was added to an energetically mixed high-temperature arrangement of high-pressure

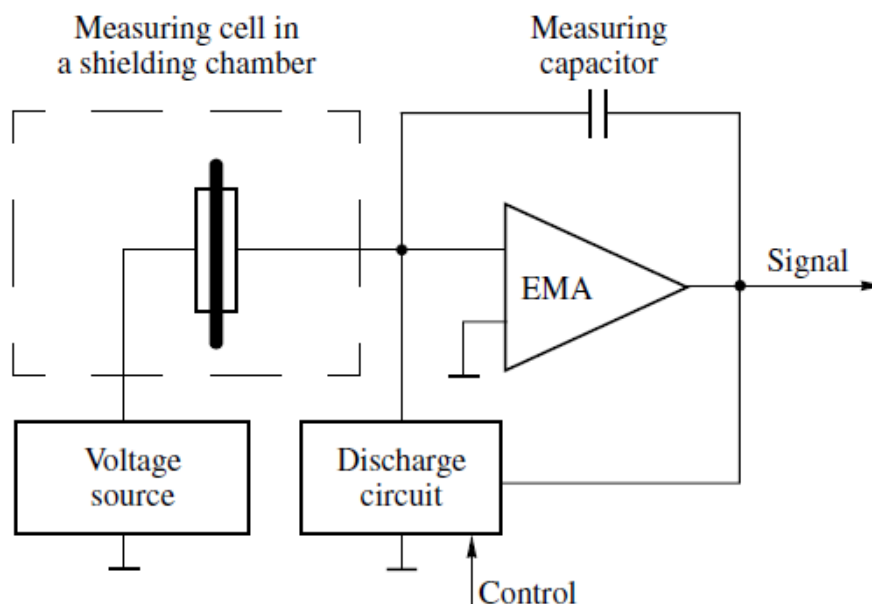


Fig. 1 Schematic of resistivity measurements.

polyethylene^{19,20}. The reactor was loaded up with argon to make a latent climate and really eliminate vaporous response items. Subsequent to washing with benzene in a Soxhlet device to eliminate the oil, the example was vacuum-dried and afterward put away in air. The resultant materials had the type of dim and dark powders^{21,22}.

The size of the metal-containing nanoparticles balanced out in a polyethylene still up in the air by transmission electron microscopy (TEM) on a JEOL JEM-100B (speeding up voltage, 75 kV). To this end, the material was sonicated in ethanol, and the scattering was applied to a copper lattice covered with poly(vinyl formal) and afterward carbon^{23,24,25}.

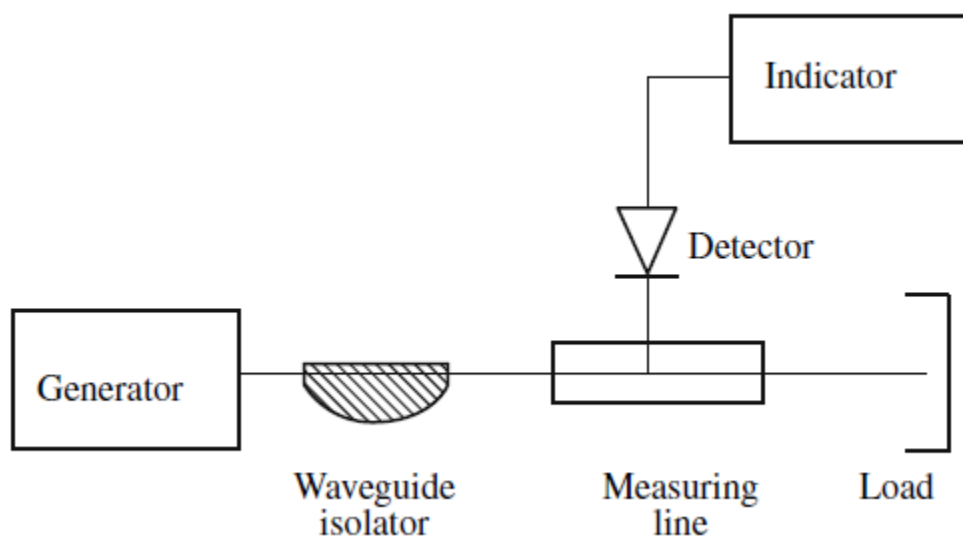


Fig. 2 Block diagram of the measuring system.

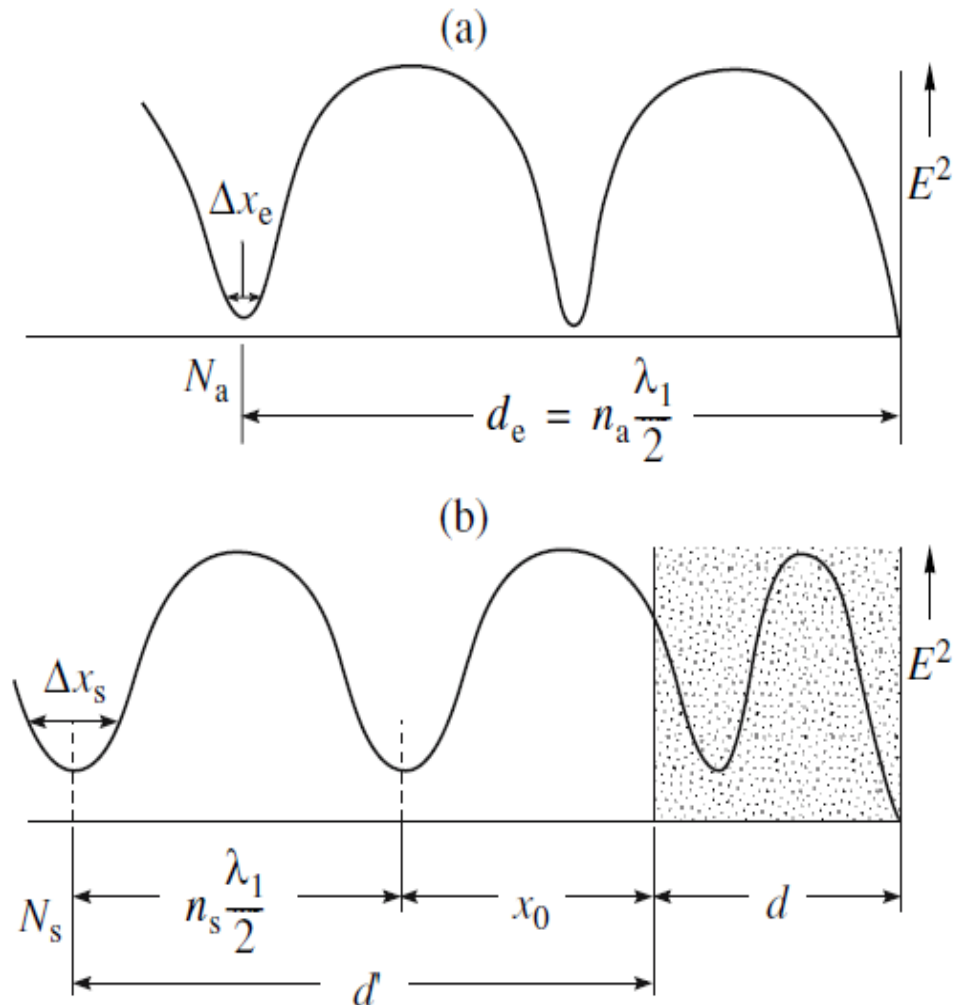


Fig. 3 Standing wave pattern in a short-circuited waveguide: (a) empty, (b) with a sample.

3 CONCLUSIONS

We arranged nanocomposites comprising of barely measured metal-containing nanoparticles implanted in a polyethylene framework and laid out conditions for the manufacture of thick movies and mass materials from the subsequent polymer powders. Our outcomes on the electrical properties of the polyethylene-lattice nanocomposites containing iron nanoparticles exhibit that, with expanding nanoparticle fixation in the polymer framework, the dielectric permittivity and retention coefficient of the nanostructured materials increment. The electrical properties (dielectric permittivity and resistivity) of the composites rely upon the nanoparticle size and fixation. Our outcomes exhibit that the microwave retention and permittivity of the nanocomposites shift minimal in an expansive recurrence range. The polarization (counting the remanent one) of the cobalt-containing nanomaterials is higher than that of the iron-containing tests. Moreover, the attractive properties of the integrated nanomaterials rely upon the convergence of metal-containing nanoparticles.

REFERENCES

1. Pomogailo, A.D., Rozenberg, A. S., and Uflyand, I. E., *Nanochastitsy metallov v polimerakh (Metal Nanoparticles in Polymers)*, Moscow: Khimiya, 2000.
2. Gubin, S.P., Koksharov, Yu. A., Khomutov, G.B., and Yurkov, G.Yu., *Magnetic Nanoparticles: Preparation, Structure, and Properties*, *Usp. Khim.*, 2005, vol. 74, no. 6, pp. 539–574.
3. Gubin, S.P., Yurkov, G.Yu., and Kosobudsky, I.D., *Nanomaterials Based on Metal-Containing Nanoparticles in Polyethylene and Other Carbon-Chain Polymers*, *Int. J. Mater. Prod. Technol.*, 2005, vol. 23, nos. 1–2, pp. 2–25.
4. Scomski, R., *Nanomagnetics*, *J. Phys.: Condens Matter*, 2003, vol. 15, pp. R841–R896.
5. Hyeon, T., *Chemical Synthesis of Magnetic Nanoparticles*, *Chem. Commun.*, 2003, pp. 927–934.
6. Lin, X.-M. and Samia, A.C.S., *Synthesis, Assembly, and Physical Properties of Magnetic Nanoparticles*, *J. Magn. Magn. Mater.*, 2006, vol. 305, pp. 100–109.
7. Kosobudsky, I.D. and Gubin, S.P., *Metallic Clusters in Polymer Matrices: A New Type of Metal-Filled Polymers*, *Vysokomol. Soedin.*, 1985, vol. 27, no. 3, pp. 689–695.
8. Smith, T.W. and Wychick, D., *Colloidal Iron Dispersions Prepared via the Polymer-Catalyzed Decomposition of Iron Pentacarbonyl*, *J. Phys. Chem.*, 1980, vol. 84, no. 12, pp. 1621–1629.
9. Gubin, S.P., *What is a Nanoparticle? Current Trends in the Development of Nanochemistry and Nanotechnology*, *Russ. Khim. Zh.*, 2020, vol. 44, no. 6, pp. 23–31.
10. Zanetti, M. and Costa, L., *Preparation and Combustion Behaviour of Polymer/Layered Silicate Nanocomposites Based upon PE and EVA*, *Polymer*, 2004, vol. 45, pp. 4367–4373.
11. Zanetti, M., Camino, G., Reichert, P., and Mülhaupt, R., *Thermal Behaviour of Poly(propylene) Layered Silicate Nanocomposites*, *Macromol. Rapid Commun.*, 2001, vol. 22, no. 3, pp. 176–180.
12. Korobov, M.S., Yurkov, G.Yu., Kozinkin, A.V., et al., *Metal-Containing Poly(tetrafluoroethylene): A Novel Material*, *Neorg. Mater.*, 2004, vol. 40, no. 1, pp. 31–40 [*Inorg. Mater. (Engl. Transl.)*, vol. 40, no. 1, pp. 26–34].
13. Yurkov, G.Yu., Baranov, D.A., Dotsenko, I.P., and Gubin, S.P., *New Magnetic Materials Based on Cobalt and Iron-Containing Nanoparticles*, *Composites*, 2006, vol. 37, no. 6, pp. 413–417.
14. Zanetti, M., Lomakina, S., and Camino, G., *Polymer Layered Silicate Nanocomposites*, *Macromol. Mater. Eng.*, 2000, vol. 279, pp. 1–9.
15. Xia, X., Cai, S., and Xie, C., *Preparation, Structure, and Thermal Stability of Cu/LDPE Nanocomposites*, *Mater. Chem. Phys.*, 2006, vol. 95, pp. 122–129.
16. Hong, J.I., Schadler, L.S., and Siegel, R.W., *Rescaled Electrical Properties of ZnO/Low Density Polyethylene Nanocomposites*, *Appl. Phys. Lett.*, 2003, vol. 82, no. 12, pp. 1956–1958.

17. Brosseau, C., Quéffélec, P., and Talbot, P., Microwave Characterization of Filled Polymers, *J. Appl. Phys.*, 2001, vol. 89, no. 8, pp. 4532–4540.
18. Ng, C.B., Ash, B.J., Schadler, L.S., and Siegel, R.W., A Study of the Mechanical and Permeability Properties of Nano- and Micron-TiO₂ Filled Epoxy Composites, *Adv. Compos. Lett.*, 2011, vol. 10, no. 3, pp. 101–111.
19. Chung, K.T., Sabo, A., and Pica, A.P., Electrical Permittivity and Conductivity of Carbon Black–Polyvinyl Chloride Composites, *J. Appl. Phys.*, 1982, vol. 53, no. 10, pp. 6867–6879.
20. Zhang, X.W., Pan, Y., Zheng, Q., et al., Time Dependence of Piezoresistance for the Conductor-Filled Polymer Composites, *J. Polym. Sci., Part B: Polym. Phys.*, 2000, vol. 38, no. 21, pp. 2739–2749.
21. Hong, J.I., Winberg, P., Schadler, L.S., and Siegel, R.W., Dielectric Properties of Zinc Oxide/Low Density Polyethylene Nanocomposites, *Mater. Lett.*, 2005, vol. 59, no. 4, pp. 473–476.
22. Cadek, M., Coleman, J.N., Barron, V., et al., Morphological and Mechanical Properties of Carbon-Nanotube- Reinforced Semicrystalline and Amorphous Polymer Composites, *Appl. Phys. Lett.*, 2002, vol. 81, no. 27, pp. 5123–5125.
23. Du, F., Scogna, R.C., Zhou, W., et al., Nanotube Networks in Polymer Nanocomposites: Rheology and Electrical Conductivity, *Macromolecules*, 2004, vol. 37, no. 24, pp. 9048–9055.
24. Hill, D.E., Lin, Y., Rao, A.M., et al., Functionalization of Carbon Nanotubes with Polystyrene, *Macromolecules*, 2002, vol. 35, no. 25, pp. 9466–9471.
25. Dang, Z.-M., Zhang, Y.-H., and Tjong, S.-C., Dependence of Dielectric Behavior on the Physical Property of Fillers in the Polymer-Matrix Composites, *Synth. Met.*, 2014, vol. 146, no. 1, pp. 79–84.